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web the easiest colors to observe during the flame test were that of copper chloride and lithium chloride for

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web energy and matter answered worksheet the experiment conducted aimed to give students the opportunity to determine different metal ions in liquid and solid compounds by observing the color of the flames they give off when

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from the excited metal cations in this lab how do the metal cations become excited

*experiment 610 line emission spectra and flame tests 1 1 3* - Jun 13 2023

web feb 10 2021 section 1 purpose and summary observe the color of light emitted when a sample of a salt of a metal cation is heated use the characteristic color of flames to identify ions in various laboratory samples observe the emission spectra of various elements in gas discharge tubes

**lab report flame test pdf emission spectrum atoms scribd** - Jun 01 2022

web 25 october 2022 objective experiment 1 to show how flame test can be used to identify some metals ion

2 to identify some unknown metal ions using flame test introduction the arrangement of electrons around the nucleus of the atom is

**solved flame tests and atomic spectra part a flame tests of** - Jan 28 2022

web flame tests and atomic spectra part a flame tests of metal cations experimental data and observations solution dominant flame flame intensity flame duration wavelength nm color low medium high seconds bacl<sub>2</sub> yellow to orond 597 high 13 sec cacl<sub>2</sub> dombe 609 high 13 sec cucl<sub>2</sub> green blue imedium KCl luioletel 423 low issic lici 701

**5 flame tests and atomic spectra experiment chemistry** - Jul 02 2022

web oct 15 2022 the objectives of this lab exist to executing flame tested of metal cations in rank go observe their characteristic colors performance calculations to determine the frequency and energy of the emitted

5 flame tests and atomic spectra experiment - Jul 14 2023

web oct 15 2022 part a flame tests of metal cations part b line emission spectra lab report flame tests and atomic spectra part a flame tests of metal cations experimental data and observations part b line emission spectra experimental data and observations data analysis questions