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 $141.9579 \text{ g/mol} \times 0.0129 \text{ m} = 2 \text{ molality}$
 $0.825 \text{ g} / 141.9579 \text{ g/mol} = 0.00581158 \text{ mol}$
 $0.00581158 \text{ mol} / 0.4500 \text{ kg} = 0.0129 \text{ m}$
 3 mole fraction $n_2 = \frac{0.825 \text{ g}}{141.9579 \text{ g/mol}} = 0.00581158 \text{ mol}$
 $n_1 = \frac{450.0 \text{ g}}{18.015 \text{ g/mol}} = 24.97918401 \text{ mol}$

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 end array

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