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together and create temporary connections that are essential to life types of chemical bonds including covalent ionic and hydrogen bonds and london dispersion forces introduction living things are made up of atoms but in most cases those atoms aren t just floating around individually

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web the relative electronegativities of the bonded atoms is h c cl and so the bond moments all point toward the cl end of the molecule and sum to yield a considerable dipole moment the molecules are relatively polar for molecules of high symmetry such as bf 3 trigonal planar ch 4 tetrahedral pf 5 trigonal bipyramidal and sf 6

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