

# Treulos Verrterisch 7 Buchstaben

Yan Bai

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electrons are shared and the bond is electronegativities equally unequally non polar polar unequally polar the degree of polarity of a bond between any two atoms is determined by consulting a table of

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web the relative electronegativities of the bonded atoms is h c cl and so the bond moments all point toward the cl end of the molecule and sum to yield a considerable dipole moment the molecules are relatively polar for molecules of high symmetry such as  $\text{BF}_3$  trigonal planar  $\text{CH}_4$  tetrahedral  $\text{PF}_5$  trigonal bipyramidal and  $\text{SF}_6$

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