

# 2 44 Oliver Street Freshwater

**JS Bruner**

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 multiple choice 2 minutes 1 pt between  
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 heat is removed at a constant rate a b  
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curve worksheet if this curve is read  
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 the diagram below illustrates the steps  
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 20 c to 10g of gaseous steam at 140 c a  
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 that is at a temperature of 30 text o  
 text c well below its meltingpoint the  
 ice is in a closed container as heat is  
 steadily added to the ice block the  
 water molecules will begin to vibrate  
 faster and faster as they absorb kinetic  
 energy

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 at a constant rate the temperature of  
 the sample is recorded at 2 minute  
 intervals in the data table below read



more heating cooling curves heating  
cooling curves page 1 heating cooling  
curves a the following graph is a  
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heat at a constant rate of 500 0 joules  
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move from gas to liquid to solid what is  
the aim of the heating curve of water  
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still use two equals n cac would be the  
specific heat of gasses water and then  
here it would be the specific heat of ice  
now again going back to the question  
we re looking for liquid water so that  
would mean the answer is option c line  
segment cd would have the specific  
heat of liquid water

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represents a sample of a substance  
starting as a solid below its melting  
point and being heated over a period of  
time which statement describes the  
energy of the particles in this sample  
during interval de a both potential  
energy and average kinetic energy  
increase

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kinetic energy of the molecules  
increases red arrows to the right  
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away from the substance dissipated to  
the surroundings and the kinetic  
energy of the molecules decreases blue

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