

Theme Of Aunt Jennifers Tigers

Mays, Kelly J.

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decomposed according to the following chemical equation
 $\text{KClO}_3 \rightarrow \text{KCl} + \text{O}_2$ molar mass of KClO_3 122.5 g/mol KCl 74.55 g/mol
 O_2 31.998 g/mol if the mass of potassium chlorate was 240
grams which of the following calculations can be used to
determine the mass of oxygen gas formed

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element in a compound is as a percentage by mass or mass
percent in part a of this lab a sample of potassium chlorate
will be experimentally analyzed in order to determine the
mass percent of

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potassium chloride or soda such as baking soda with water
mixing the two creates a chemical reaction that uses up
energy which makes the mixture colder you can see which
substance cools off water the most by trying out this
experiment

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web the potassium chloride to the water make sure all of the
 potassium chloride is emptied into the water data will be
 collected for 180 seconds 5 using a glass stirring rod gently
 stir the water in the beaker for about 20 seconds to help the
 potassium chloride dissolve 6 after 180 seconds have lapsed
 the calculator

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sodium chloride has been known since ancient times
archaeologists have found evidence that rock salt was used
as early as 3000 bc and brine as early as 6000

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potassium chlorate is produced by passing chlorine into hot
milk of lime and subsequently adding potassium chloride
liebig process the electrolysis of kcl in aqueous solution is
also used sometimes in which the chloride ions formed at the
anode react with koh in situ

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web once all the magnesium ribbon has reacted discard the
mixture in the sink with plenty of water rinse out and dry the

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sodium chloride water neutralisation $\text{NaOH (aq) + HCl (aq) \rightarrow NaCl (aq) + H}_2\text{O (l)}$