

# Change Happens When The Pain Of Staying The Same

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imagine that you have a block of ice  
that is at a temperature of 30 c well  
below its melting point the ice is in a  
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added to the ice block the water  
molecules will begin to vibrate faster  
and faster as they absorb kinetic  
energy

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component of phases and phase  
changes explaining what equations to  
use what specific heat is

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temperature versus cooling time in  
figure pageindex 4 plots temperature  
versus time as a 75 g sample of steam  
initially at 1 atm and 200 c is cooled  
although we might expect the cooling  
curve to be the mirror image of the  
heating curve in figure pageindex 3 the  
cooling curve is not an

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web once all of the material has  
solidified at the time indicated by point  
c the cooling will continue at a rate  
determined by the heat capacities of  
the two solids a and b the composition  
and of course the geometry of the  
experimental set up by measuring  
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composition one can map the entire  
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graphs they plot a substance s  
temperature y axis against heat x axis  
for heating curves we start with a solid  
and add heat energy for cooling curves  
we start with the gas phase and remove  
heat energy cooling and heating curves  
have five segments let s study each one  
by looking at heating ice to

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because the boiling point for water is  
100 c different substances have  
different melting points and boiling  
points but the shapes of their heating  
curves are very similar for example this  
is the heating curve for iron a metal  
that melts at 1538 c and boils at 2861 c  
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