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 $6.941 \text{ amu} \times 6.0151 \text{ amu} \times 1 \times 7.0160 \text{ amu}$   
 $6.941 \times 1.0009 \times 7.0160 \times 1.0009 \times 0.075 \times 0.075 \times 0.075$  corresponds to a  
 natural abundance of 6li of 7.5 percent  
 the natural abundance of 7li is  $\times 0.925$   
 or 92.5 percent 3.7 the unit factor  
 required is 6.022  
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 $01 \text{ g}$  3  $16.00 \text{ g}$  73.89 g b molar mass  
 $\text{CS}_2$  12  $01 \text{ g}$  2  $32.07 \text{ g}$  76.15 g c molar  
 mass  $\text{CHCl}_3$  12  $01 \text{ g}$  1  $008 \text{ g}$  3  $35.45 \text{ g}$   
 $119.37 \text{ g}$  d molar mass  $\text{C}_6\text{H}_8\text{O}_6$  6  $12.01 \text{ g}$   
 $8.1008 \text{ g}$  6  $16.00 \text{ g}$  176.12 g e molar  
 mass  $\text{KNO}_3$  39  $10 \text{ g}$  14  $01 \text{ g}$  3  $16.00 \text{ g}$   
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