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 balancing 1 Na 3 PO_4 3 KOH 3 NaOH 1 K 3 PO_4 1 MgF_2 1 Li 2 CO_3 1 MgCO_3 2 LiF 1 P 4 3 O 2 2 P 2 O 3 2 RbNO_3 1 BeF_2 1 Be NO_3 2 2 RbF

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web the first step to balance the equation is to write down the chemical formula of reactants that are listed on the left side of the chemical equation after this you can list down the

products on the right hand side of the chemical equation
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 $3\text{ KClO}_3 \rightarrow 3\text{ KCl} + 2\text{ O}_2$
 $2\text{ NaCl} + \text{F}_2 \rightarrow 2\text{ NaF} + \text{Cl}_2$
 $2\text{ H}_2 + \text{O}_2 \rightarrow 2\text{ H}_2\text{O}$
 $5\text{ Pb} + 2\text{ OH}^- + 2\text{ HCl} \rightarrow 2\text{ H}_2\text{O} + 1\text{ PbCl}_2 + 2\text{ AlBr}_3 + 3\text{ K}_2\text{SO}_4 + 6\text{ KBr} + 1\text{ Al}_2\text{SO}_4 + 3\text{ CH}_4 + 2\text{ O}_2 + 1\text{ CO}_2 + 2\text{ H}_2\text{O} + 8\text{ H}^+ + 3\text{ C}_3\text{H}_8 + 5\text{ O}_2 + 3\text{ CO}_2 + 4\text{ H}_2\text{O}$

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 $1\text{ O}_2 + 2\text{ NH}_3 \rightarrow 3\text{ H}_2\text{O} + 2\text{ H}_2$
 $2\text{ O}_2 + 3\text{ H}_2 \rightarrow 2\text{ H}_2\text{O} + \text{O}_2$
 $4\text{ RbH} + 2\text{ O} \rightarrow 2\text{ RbOH} + 2\text{ H}_2$
 $5\text{ Cl}_2 + 2\text{ KBr} \rightarrow 2\text{ KCl} + \text{Br}_2$
 $6\text{ N}_2 + 2\text{ H}_2 + 3\text{ NH}_3 + 7\text{ N}_2\text{F}_6 + 3\text{ Ca} + \text{N}_2\text{O} + 2\text{ C}_2\text{H}_5\text{OH} + 2\text{ H}_2\text{O} + 9\text{ C}_2\text{H}_5\text{OH}$

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 $\text{F}_2 + 2\text{ NaCl} \rightarrow 2\text{ NaF} + \text{Cl}_2$
 $2\text{ H}_2 + \text{O}_2 \rightarrow 2\text{ H}_2\text{O}$
 $\text{Pb} + \text{OH}^- + 2\text{ HCl} \rightarrow 2\text{ H}_2\text{O} + 1\text{ PbCl}_2 + 2\text{ AlBr}_3 + 3\text{ K}_2\text{SO}_4 + 6\text{ KBr} + 1\text{ Al}_2\text{SO}_4 + 3\text{ CH}_4 + 2\text{ O}_2 + 1\text{ CO}_2 + 2\text{ H}_2\text{O}$

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 $4\text{ Fe} + 3\text{ O}_2 \rightarrow 2\text{ Fe}_2\text{O}_3$
 $2\text{ FeBr}_3 + 3\text{ H}_2\text{SO}_4 \rightarrow 1\text{ Fe}_2(\text{SO}_4)_3 + 3\text{ H}_2\text{O} + 2\text{ C}_2\text{H}_5\text{OH}$

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 $38\text{ Fe} + 2\text{ O}_2 + 3\text{ CO} + 2\text{ Fe}_3\text{CO}_2 + 39\text{ H}_2 + 3\text{ PO}_4 + 3\text{ BaO} + \text{Ba} + 3\text{ PO}_4 + 2\text{ H}_2 + \text{O}_2 + 40\text{ Mg} + \text{NH}_4 + \text{PO}_4 + \text{Mg} + 2\text{ P}_2 + \text{O}_2 + 7\text{ NH}_3 + \text{H}_2 + \text{O}_2 + 41\text{ Al} + 3\text{ H}_2 + \text{SO}_4 + \text{Al}_2 + \text{SO}_4 + 3\text{ H}_2 + 42\text{ P}_4 + \text{O}_2 + 10\text{ HCl} + \text{POCl}_3 + 3\text{ HPO}_3 + 43\text{ H}_2\text{O} + 2\text{ H}_2 + \text{O}_2 + 3\text{O}_2 + 4\text{ HNO}_2$

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