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calculate the average atomic mass of
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purpose this lab exercise is designed to
show you isotopes of an element in a
simulation form you will be asked to
gather data about the isotopes and
organize the data if atoms were as
large as beans they could be sorted
counted and massed

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obtained a sample of beanium a

fictitious element in order to learn about isotopes and average atomic mass beanium has three isotopes white brown and speckled which have different masses and different percent abundances

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web 1 obtain a sample of beanium separate the pintonium navium and lentilium and then mass each group record the values in the data table 2 count the number of each isotope by counting the pintonium navium and lentilium 3 calculate the total number of isotopes

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