

# Is Haunting Adeline A Series

Hussin A.Rothana

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 $2 \text{ CO}_2(g) + 2 \text{ H}_2\text{O}(g) \rightarrow \text{C}_2\text{H}_2(g) + 5 \text{ O}_2(g)$   $\Delta h = 2 \text{ H}_2(g) + \text{C}_2\text{H}_6(g) \rightarrow \Delta h = 94.5 \text{ kJ}$   
 $\text{H}_2\text{O}(g) + \text{H}_2(g) + \frac{1}{2} \text{ O}_2(g) \rightarrow \Delta h = 71.2 \text{ kJ}$   
 $\text{C}_2\text{H}_6(g) + \frac{7}{2} \text{ O}_2(g) \rightarrow 2 \text{ CO}_2(g) + 3 \text{ H}_2\text{O}(g)$   $\Delta h = 283 \text{ kJ}$   
 $\text{C}_2\text{H}_6(g)$

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web thus for reaction 9 2 in which 1 mole of  $\text{H}_2\text{SO}_4$  and 1 mole of  $\text{OH}^-$  combine to form 2 moles of  $\text{H}_2\text{O}$   $\Delta H = 55.8 \text{ kJ}$  as shown note that the enthalpy change  $\Delta H_{\text{products}} - \Delta H_{\text{reactants}}$  is positive if

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