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this reaction is $\text{Mg} + 2\text{HNO}_3 \rightarrow \text{Mg}(\text{NO}_3)_2 + \text{H}_2$ g in this equation the s

symbolizes solid magnesium aq

represents aqueous nitric acid aq

signifies magnesium nitrate in a

dissolved state and g represents the

gaseous hydrogen produced

$\text{HNO}_3 + \text{Mg} + 2\text{H}_2\text{SO}_4 \rightarrow \text{H}_2\text{SO}_4 + \text{Mg}(\text{NO}_3)_2$ chemical equation - Aug 11 2023

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magnizium in elemental form h_2 there are 2 molecules of nitric acid each acid has one hydrogen so two molecules of nitric acid is 2 hydrogens n_2 there are 2 molecules of nitric acid each acid has one nitrogen so two molecules of nitric acid is 2 nitrogens o_6

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