

Oh Happy Day Partition

J Elliott

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 $\text{Na}_2\text{CO}_3(\text{aq}) + 2\text{AgNO}_3(\text{aq}) \rightarrow \text{Ag}_2\text{CO}_3(\text{s}) + 2\text{NaNO}_3(\text{aq})$ The net ionic equation is: $2\text{Ag}^+(\text{aq}) + \text{CO}_3^{2-}(\text{aq}) \rightarrow \text{Ag}_2\text{CO}_3(\text{s})$
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 $\text{Ca}^{2+}(\text{aq}) + \text{H}_2\text{CO}_3(\text{aq}) \rightarrow \text{CaCO}_3(\text{s}) + \text{H}_2\text{O}(\text{l})$ May be wrong, but it seems reasonable. The gas would interact ... If the pH=9 for a solution of KOH(aq), what is the ... - Socratic Aug 18, 2016 · If the pH = 9 for a solution of KOH (aq), what is the concentration of H_2O ?
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