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curve is a plot of the ph of a solution as

a function of the amount of base usually

in ml added to an acidic solution or as

the amount of acid added to a boss

solution the base is added to

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web where the salt is the conjugate base of the acid the four parts of the titration curve are described below and you should look to the appropriate text

section to see how they are treated pure acid 0 mL of base is added section 17.3.2.1

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web at the equivalence point in an acid base titration moles of base moles of acid and the solution only contains salt and water acid base titrations are monitored by the change of pH as

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purpose the purpose of this lab is to determine the shape of acid base titration curves theoretically and experimentally and to use an acid base titration to measure the effectiveness of antacid tablets summary a titration curve is the plot of the ph value of a solution vs the amount of titrant added *acid base titrations chemistry libretexts* - Feb 28 2023

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web ph titration graph 1 lab data provide columns a and b 2 a graph of ph column b plotted as ordinate vs volume column a as abscissa provides a titration curve as illustrated by graph 1

or 2 below 3 estimate of the midpoint of the steeply rising section of the titration curve provides an evaluation of the equivalence point

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base titration curve from titration with

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added ml ph of the solution 4 5 4 6 4 6

5 9 5 9

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web background plot of the ph of a solution against the volume of titrant added is called a titration curve the ph can be measured directly with a ph meter while titrant is added from a buret for the acids used in this experiment the titrant will always be a

0 1 m solution of the strong base naoh

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started with a slow gradual change in ph as it reached the equivalence point
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