

## S Based On Songs

# Tao Wei

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web solution a titrant volume 0 ml the solution ph is due to the acid ionization of hcl because this is a strong acid the ionization is complete and the hydronium ion molarity is 0 100 m the ph of the solution is then  $\text{ph} = -\log 0.100 = 1.000$  ph  $\log 0.100 = 1.000$  b titrant volume 12 50 ml

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from a buret for the acids used in this experiment the titrant will always be a 0.1 M solution of the strong base NaOH

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