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the two atoms in the bond is unequal a
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where the bond polarities do not cancel

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- Feb 09 2023 web the relative electronegativities of the bonded atoms is h c cl and so the bond moments all point toward the cl end of the molecule and sum to yield a considerable dipole moment the molecules are relatively polar for molecules of high symmetry such as BF_3 trigonal planar CH_4 tetrahedral PF_5

trigonal bipyramidal and SF_6

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