

# Smoke On The Water Piano

**Judd E. Hollander**

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 web 11 12 heating and cooling curves when a substance isolated from its environment is subjected to heat changes corresponding changes in temperature and phase of the substance is observed this is graphically represented by heating and cooling curves

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web determine the final temperature and state of the ammonia once the system reaches equilibrium assuming that no heat is lost to the surroundings use the information below  
 normal boiling point of  $\text{NH}_3$  33.3 °C  $\Delta H_{\text{vap}}$  23.4 kJ/mol  
 specific heat  $\text{NH}_3$  4.74 J/g °C specific heat  $\text{NH}_3$  2.06 J/g °C  
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 figure 11.16 11.16.1 in the heating curve of water the temperature is shown as heat is continually added changes of state occur during plateaus because the temperature is constant

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web once all of the material has solidified at the time indicated by point c the cooling will continue at a rate determined by the heat capacities of the two solids a and b the composition and of course the geometry of the

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web for water this temperature is 100 c because the boiling point for water is 100 c different substances have different melting points and boiling points but the shapes of their heating curves are very similar for example this is the heating curve for iron a metal that melts at 1538 c and boils at 2861 c cooling curves

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