

Michael Hirte Neue Freundin Sandra

WJ Hussar

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signifies solutions labelled with molar concentration a 1 0 m
solution contains 1 mole of solvent per litre of solution

molality is the number of solvent moles per kilogram
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9579 g mol x 0 0129 m 2 molality 0 825 g 141 9579 g mol 0
00581158 mol 0 00581158 mol 0 4500 kg 0 0129 m 3 mole
fraction na 2 hpo 4 0 825 g 141 9579 g mol 0 00581158 mol
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4 2 2 a first convert 250 ml to liters 250 1000 0 25 then
calculate molarity 5 moles 0 25 liters 20 m 3 c a solution
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480 520 720 624 1000 1 344 m q5 with increase in
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