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a function of the amount of base usually

in ml added to an acidic solution or as

the amount of acid added to a boss

solution the base is added to

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bases the shape of the titration curve

for a weak acid or base depends on the

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naoh added to 10 00 ml of 1 00 m

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word titulus which means inscription or

title

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web where the salt is the conjugate

base of the acid the four parts of the

titration curve are described below and

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section to see how they are treated

pure acid 0 ml of base is added section

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added ml ph of the solution 4 5 4 6 4 6

5 9 5 9

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added is called a titration curve the ph

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completely ionize in solution resulting

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there s only partial ionization the

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acid showed a stretched out curve as it

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