

# Warren Buffett List

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chemistry that deals with the relations  
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electrical energy because an electrical  
current is a flow of electrical charges  
electrochemistry is primarily concerned  
with electron transfer reactions which  
are based on oxidation reduction  
phenomena

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and there is no formal lab report  
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of elements ordered by their standard  
electrode potentials it tells us how  
easily each element is oxidised  
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involving the transfer of electrons from  
one reactant to another are called  
oxidation reduction reactions or redox  
reactions in a redox reaction two half  
reactions occur one reactant gives up  
electrons undergoes oxidation and  
another reactant gains electrons  
undergoes reduction

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web an experiment that will illuminate the mysteries of ions create a cell of electrolyte and metals and watch how the reactive and non reactive metals form ion this practical helps students to grasp the ideas around electromotive force found in chemicals this experiment should take 45 minutes

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