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nonpolar hcn h c n cf<sub>4</sub> c f<sub>4</sub> cocl<sub>2</sub> c o c l  
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2 answer problem 6 2 3 6 2 3

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web the relative electronegativities of the bonded atoms is h c cl and so the bond moments all point toward the cl end of the molecule and sum to yield a considerable dipole moment the molecules are relatively polar for molecules of high symmetry such as bf<sub>3</sub> trigonal planar ch<sub>4</sub> tetrahedral pf<sub>5</sub> trigonal bipyramidal and sf<sub>6</sub>

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