

# Yarc Secondary Conversion Tool

S Ben Porath

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which is heaviest a 25 g of hg b 2 mole of h<sub>2</sub>o c 2 mole of co<sub>2</sub> d 4 g atom of o q

2 16 g of sox occupies 5 6 litre at stp assuming ideal gas nature the value of

x is a 1 b 2 c 3 d none of these q 3 the density of liquid mol wt 70 is 1 2 g ml 1

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02x10<sup>23</sup> 5 hint first find the number of moles of oxygen used 1027 8g 1023 2g

4 6g o<sub>2</sub> 2 16 32 g mol n m mr n 4 6 32 n 0 144 mol y n l y 0 144 6 02 1023 y 8

669 molecules

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100h<sub>2</sub>o 100 h 2 50 o 2 100 h 2 o

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how many atoms are contained in the following a 1 molecule of  $\text{CH}_3\text{CO}_2\text{H}$  b 15 molecules of  $\text{NH}_4\text{Cl}$  c 2.56 mol of  $\text{NH}_4\text{PO}_4$  d 0.0871 mol of  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$  11 find the mass in grams of each of the following a 1 pb atom b 235 ag atoms c 4.51 x 10<sup>22</sup> h 2 o

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web the amount in moles of pb 3 o 4 to 3 significant figures iii the amount in moles of lead produced iv mass of lead produced to 3 significant figures b 35

61 g of tin reacts with exactly 42.60 g of chlorine  $\text{Cl}_2$  to form 78.21 g of tin iv chloride  $\text{SnCl}_4$  i calculate the number of moles of tin chlorine and tin chloride a r values

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