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Marcel A. Müller

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point of an acid base titration is the
point at which exactly enough acid or
base has been added to react
completely with the other component
the equivalence point in the titration of
a strong acid or a strong base occurs at
ph 7 0

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base phenolphthalein changes color in
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objectives describe a titration
experiment explain what an indicator
does perform a titration calculation
correctly the reaction of an acid with a
base to make a salt and water is a
common reaction in the laboratory
partly because so many compounds can
act as acids or bases

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are those of a strong acid with a strong
base table 15 7 1 shows data for the
titration of a 25 0 ml sample of 0 100 m
hydrochloric acid with 0 100 m sodium
hydroxide the values of the ph
measured after successive additions of
small amounts of naoh are listed in the
first column of this table and are
graphed in

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reaction between an acid and a base can be very useful if an acidic solution of known concentration a standard solution is added to a basic alkaline solution of unknown concentration until the solution is exactly neutralised i.e. there is only salt and water it is possible to calculate

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web other articles where acid base titration is discussed titration the sample and titrant include acid base titrations precipitation titrations complex formation titrations and oxidation reduction redox titrations in acid base titration i.e. the titration of an acid with a base or vice versa the indicator is a substance that can exist in two forms an

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web the above equation works only for neutralizations in which there is a 1:1 ratio between the acid and the base the example below demonstrates the technique to solve a titration problem for a titration of sulfuric acid with sodium hydroxide

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help of a standard solution which could be of either base or acid of a known concentration

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titration curves equivalence point

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web titrant and analyte is a pair of acid and base acid base titrations are monitored by the change of pH as titration progresses let us be clear about some terminologies before we get into the discussion of titration curves titrant solution of a known concentration which is added to another solution whose concentration has to be determined

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with an acid

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