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first one is done for you 23 $\text{Fe}^{2+} \rightarrow \text{Fe}^{3+} + \text{e}^{-}$ oxidation $\text{Co}^{2+} + 2\text{e}^{-} \rightarrow \text{Co}$ reduction $\text{Fe}^{2+} + 2\text{e}^{-} \rightarrow \text{Fe}$ 24 3 $\text{Ag}^{+} + \text{e}^{-} \rightarrow \text{Ag}$ oxidation reduction 25 $\text{Cu}^{2+} + 2\text{e}^{-} \rightarrow \text{Cu}$ oxidation reduction 26 $\text{O}_2 + 4\text{e}^{-} \rightarrow 2\text{O}^{2-}$

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