

# Animales En Peligro De Extincion En Tabasco

Terry C. Jones

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10 4s 2 4p 6 4d 10 5s 2 5p 5

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web 1 the aufbau principle states that  
electrons are always placed in the  
lowest energy sublevel that is available  
2 the pauli exclusion principle states  
that each orbital may contain a  
maximum of 2 electrons electrons must  
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minimum of 6 electrons each with  
opposite spins an atomic orbital can  
hold a maximum of 6 electrons each  
with the same spin an atomic orbital  
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with opposite spins multiple choice edit  
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web the four different types of orbitals s p d and f have different shapes and one orbital can hold a maximum of two electrons the p d and f orbitals have different sublevels thus can hold more electrons as stated the electron configuration of each element is unique to its position on the periodic table

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is the noble gas helium which has an atomic number of 2 the helium atom contains two protons and two electrons the first electron has the same four quantum numbers as the hydrogen atom electron n 1 l 0 ml 0 m s 1 2

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