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Química Ronald J. Gillespie,1988-08

Principios de química Richard E. Dickerson,1992-11 Esta tercera edición de Principios de Química, se ha proyectado para ser utilizada en un curso universitario de Química general, que debe servir tanto como una visión de la Química para los especialistas, como de una buena base para los estudios posteriores de las disciplinas de Química. Por consiguiente, hay varios capítulos que pretenden introducir las diferentes áreas de la Química, incluida la inorgánica, nuclear, orgánica y Bioquímica, y se intenta, a lo largo de todo el libro, colocar a la Química en su marco histórico y cultural. Al mismo tiempo, se presentan los aspectos cuantitativos de la Química de forma consecuente con su importancia, de manera que resulte fácil apoyarse en ellos en los cursos posteriores.

Fundamentos de Química Ralph A. Burns,2003

Química general Gordon M. Barrow,1975-11 Sin introducir grandes modificaciones en los temas que habitualmente se encuentran en un texto de esta clase, el autor ha tratado de mostrar que la Química se origina para proporcionar una perspectiva adecuada y entrar en el tema.

Química inorgánica Therald Moeller,1981-02

Conceptos química inorgánica Bodie Eugene Douglas,John J. Alexander,1987 Este texto debe ayudar a los estudiantes a integrar sus conocimientos de Química, capacitándolos para aprovechar el caudal de conocimientos adquiridos en cursos de Química independientes.Desde la primera edición de este libro, los avances de la Química inorgánica han sido impresionantes. Para mantener el texto al día se han hecho las adiciones correspondientes y necesarias. Se han utilizado figuras más profusamente que en la primera edición y, asimismo, se han seleccionado cuidadosamente para que resulten más efectivas.

Química Páez Bautista Cárdenas,2012-01-01 Esta es una obra que propone desarrollar los contenidos básicos del curso de química general, mostrando su relación con los aspectos de la salud humana y la sostenibilidad del ambiente. Para esto, se desarrollan cuatro grandes temas: • Módulo 1. Química: la ciencia de las sustancias. • Módulo 2. Identidad y transformación de las sustancias. • Módulo 3. Gases y disoluciones. • Módulo 4. Química del carbono. En cada uno de los cuatro módulos se incluyen ejemplos y ejercicios de aplicación, lecturas sobre avances de la ciencia y la tecnología y su impacto en la salud y el

ambiente (CTSA), cuatro talleres de aprendizaje cooperativo y una evaluación que promueve el manejo apropiado de conceptos, así como las competencias de indagación, manejo de diferentes fuentes de información, argumentación y, comunicación oral y escrita.

Fundamentos de la química general e inorgánica Hans Rudolf Christen,1977 Este texto tiene como objetivo proporcionar a los estudiantes una introducción en los fundamentos teóricos de nuestra ciencia que les permita alcanzar simultáneamente una primera versión general sobre el campo de la Química inorgánica. De acuerdo con esta idea, la parte general -teoría atómica y de enlaces, termodinámica, cinética, reacciones ácido-base y redox, Química de los complejos- ocupa un espacio bastante amplio, mientras que la Química descriptiva -campo del estudio de las sustancias propiamente dicho- queda más bien en un segundo término sin formar el objetivo fundamental de la obra.

Química de coordinación de iones metálicos en estado de oxidación II derivados de alfa-hidroxicarboxilatos. ,

Química inorgánica Enrique Gutiérrez Ríos,2024-10-03 La finalidad de este libro es ofrecer un panorama general de la Química inorgánica a alumnos universitarios de Química. En su orientación se aparta, en algunos aspectos, de lo habitual en tratados de esta clase. Estos cambios se basan en motivos doctrinales y docentes.

Química inorgánica Alan G. Sharpe,2023-10-26 Este libro no es un texto de Química inorgánica industrial. Su objetivo es proporcionar una introducción crítica a la Química inorgánica moderna. Se ha intentado que sea claro y de fácil lectura, orientado más al estudiante que al profesor. Para facilitar la comprensión de las materias estudiadas, al final de cada capítulo se incluyen algunos problemas y no de meras cuestiones de revisión. A lo largo de todo el texto se ha tenido cuidado en distinguir las magnitudes de definición exacta, de las que aún teniendo una definición precisa, no se pueden medir sin la introducción de ciertos supuestos.

Nuevos complejos organometálicos de titanio y circonio en estado de oxidación IV María José Ruiz García,1997
Elementos de Transición ,

Química general. Introducción a la Química Teórica Cristóbal Valenzuela Calahorro,1995-01-01

Agrochimica ,1995

Química: la Ciencia Central Theodore L. Brown,H. Eugene LeMay,Bruce Edward Bursten,Julia R. Burdge,2004 The scientific accuracy, clarity, and visuals of Chemistry: The Central Science make it the most trusted chemistry book available. A comprehensive media package works in tandem with the text. Translated into Spanish.Introduction: Matter and Measurement, Atoms, Molecules, and Ions, Stoichiometry: Calculations with Chemical Formulas and Equations, Aqueous Reactions and Solution Stoichiometry, Thermochemistry, Electronic Structure of Atoms, Periodic Properties of the Elements, Basic Concepts of Chemical Bonding, Molecular Geometry and Bonding Theories, Gases, Intermolecular Forces, Liquids, and Solids, Modern Materials, Properties of Solutions, Chemical Kinetics, Chemical Equilibrium, Acid-Base Equilibria, Additional

Aspects of Aqueous Equilibria, Chemistry of the Environment, Chemical Thermodynamics, Electrochemistry, Nuclear Chemistry, Chemistry of the Nonmetals, Metals and Metallurgy, Chemistry of Coordination Compounds, The Chemistry of Life: Organic and Biological Chemistry. For anyone interested in learning more about chemistry.

Química G. William Daub, William S. Seese, 1996

Química orgánica simplificada Rudolph Macy, 1976 Se dedica especialmente este libro a dos grupos de lectores: a aquellos que, ocupados en trabajos de carácter químico, no están en disposición de recibir un curso normal de Química orgánica y a los que, asistiendo a un curso regular, sienten la necesidad de un libro complementario que les ayude adecuadamente a acercarse a la Química orgánica.

Principios básicos de química Harry B. Gray, Gilbert P. Haight, 1975 Este texto es una respuesta al reto que significa presentar una relación innovadora y moderna de los principios de la Química. El rápido desarrollo de los últimos cursos preuniversitarios está creando presiones sin precedentes que obligan a cambiar la enseñanza en el curso de iniciación. En este nivel se necesitan urgentemente aproximaciones al estudio de los campos más avanzados de la Química, con una presentación puesta al día. Además del texto propiamente dicho, existe la Guía para el profesor.

Bioquímica Jeremy Mark Berg, Lubert Stryer, John L. Tymoczko, 2007 Esta nueva edición del Stryer mantiene todas las características que han hecho de esta obra un auténtico best-seller en la enseñanza de la materia. Las razones de este éxito son diversas: texto claro y riguroso, contenido amplio y actualizado, ilustrado, ilustraciones didácticas y atractivas... En esta sexta edición se han enriquecido y actualizado sus contenidos, destacando las siguientes aportaciones: se pone un mayor énfasis en los aspectos fisiológicos, se ofrece una perspectiva molecular de la evolución y se incluyen nuevos capítulos sobre la hemoglobina y el desarrollo de fármacos, así como nuevas aplicaciones clínicas y mejores visualizaciones de las estructuras moleculares. Es de destacar el material complementario incluido en la espectacular página web del libro (www.whfreeman.com/stryer y próximamente en www.reverte.com/stryer), muy útil para facilitar el proceso de aprendizaje.

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