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Administración - Qué es, concepto, objetivos, tipos, historia Equipo editorial, Etecé (16 de abril de 2025). Administración. Enciclopedia Concepto. Recuperado el 18 de agosto de 2025 de <https://concepto.de/administracion/>. **¿Qué es la Administración? Definición, Concepto y Ejemplos ...** Jul 24, 2025 · ¿Qué es Administración de Empresas? Cuando la gente pregunta 'qué es administración', casi siempre tiene en la cabeza una empresa. La administración de empresas es, ni más ni menos, aplicar todo el proceso administrativo (planear, organizar, dirigir y controlar) en una organización privada que busca generar ganancias. **Administración - Qué es, definición y concepto** Publicado por Julián Pérez Porto y Ana Gardey. Actualizado el 25 de julio de 2023. Administración - Qué es, definición y concepto. Disponible en <https://definicion.de/administracion/> **¿Qué es la administración? Para qué sirve, tipos y objetivos** Apr 3, 2025 · Descubre como la Administración puede ser utilizada por instituciones y organizaciones para alcanzar sus metas. Conoce los fundamentos de la Administración. **ADMINISTRACIÓN in English - Cambridge Dictionary** ADMINISTRACIÓN translations: administration, management, agency, administration, manager's office, administering.... Learn more in the Cambridge Spanish-English ... **Administración - Wikipedia, la enciclopedia libre** Organización industrial del trabajo. La administración es una de las « actividades humanas » más importantes encargada de organizar y dirigir el trabajo individual y colectivo efectivo en términos de objetivos. A medida que la sociedad comenzó a depender crecientemente del esfuerzo grupal y que muchos grupos organizados tienden a crecer, la tarea de los administradores ... **Administración. Qué es, significado, importancia y objetivos** GestioPolis.com Experto. (2020, octubre 6). Administración. Qué es, definición, significado, importancia y objetivos. Recuperado de <https://www.gestiopolis.com/administracion/>

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 $2 \text{ CO}_2 \text{ g} + \text{H}_2\text{O g} \rightarrow \text{C}_2\text{H}_2 \text{ g} + 2 \text{O}_2 \text{ g}$ $\Delta h = 2 \text{ h}_2 \text{ g} + \text{C}_2\text{H}_6 \text{ g}$ $\Delta h = 94.5 \text{ kJ}$
 $\text{H}_2\text{O g} + \text{H}_2 \text{ g} + \frac{1}{2} \text{O}_2 \text{ g} \rightarrow \text{C}_2\text{H}_6 \text{ g}$ $\Delta h = 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6 \text{ g} + 2 \text{ CO}_2 \text{ g} + 3 \text{ H}_2 \text{ o g} \rightarrow \Delta h = 283 \text{ kJ}$ $\text{g} + 1 \text{ C}_2\text{H}_6 \text{ g}$

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$\text{H}_4 \text{ g} + 3 \text{ O}_2 \text{ g} \rightarrow 2 \text{ CO}_2 \text{ g} + 2 \text{ H}_2 \text{ O l}$ $\Delta h = 1411 \text{ kJ}$
 $\text{mole C}_2\text{H}_6 \text{ g} + 7 \text{ O}_2 \text{ g} \rightarrow 2 \text{ CO}_2 \text{ g} + 3 \text{ H}_2 \text{ O l}$ $\Delta h = 1560 \text{ kJ}$
 $\text{mole H}_2 \text{ g} + \frac{1}{2} \text{ O}_2 \text{ g} \rightarrow \text{H}_2 \text{ O l}$ $\Delta h = 285 \text{ kJ}$ mole

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o₂ g 2 co₂ g 2 h₂o l h 1411 kj c₂h₆ g 3½ o₂ g 2 co₂ g 3 h₂o l
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given trial 1 use hess s law and the following equations and
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2 h 4

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web thus for reaction 9 2 in which 1 mole of h₃o and 1 mole
of oh combine to form 2 moles of h₂o h 55 8 kj as shown note
that the enthalpy change h h_{products} h_{reactants} is positive if
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