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Sabine Zange

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web the initial rate law is then measured for each of the reactions consider the reaction between nitrogen monoxide

gas and hydrogen gas to form nitrogen gas and water vapor
the following data were collected for this reaction at 1280
text o text c see table below notice that the starting
concentrations of ce no and ce h

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web determination of a rate law and activation energy in this
experiment glass beakers pipettes and a hotplate were
utilized to set up the procedures a solution of deionized
water a buffer potassium iodide ki sodium thiosulfate
na₂s₂o₃ starch and hydrogen peroxide h₂o₂ were used in
order to calculate the rate of reaction

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equations are mathematical expressions that describe the
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reaction described by the chemical equation a a b b p r o d u
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thinks fit order that there shall be included in the sum for
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empirical equation that adequately describes reaction rate

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