

Oh But My Darling What If You Fly

Jiyuan Zhang

What is the chemical formula for lithium hydroxide? - Socratic Apr 15, 2017 · Lithium is a Group 1 metal and commonly forms a M^+ ion. Hydroxide anion, OH^- , has a unit negative charge. When they make music together, there is thus 1:1 stoichiometry between ions: $Li(s) + H_2O(l) \rightarrow LiOH(aq)$

What is the product of the following reaction?
1) $CH_3OH + NH_3(aq) \rightarrow$ - Socratic Dec 23, 2017 · These are ostensibly acid-base reactions.... For ammonium we could write.... $NH_4^+ + OH^- \rightarrow NH_3(aq) + H_2O(l)$ For methanol, the acid base reaction would proceed ... Could a buffered solution be made by mixing aqueous ... - Socratic Mar 25, 2017 · Could a buffered solution be made by mixing aqueous solutions of HCl and NaOH? Explain. Why isn't a mixture of a strong acid and its conjugate base considered a buffered solution?

0.10 moles of sodium cyanate dissolved in 250 cm³ distilled Dec 28, 2015 · The cyanate anions will react with water to form isocyanic acid, HNCO, and hydroxide anions, OH^- . This tells you that you can expect the pH of the resulting solution to be higher than 7. Calculate the molarity of the sodium cyanate solution by using the given number of moles and volume. Keep in mind that you need to use liters of solution and ...

What is the formula of "magnesium hydroxide"? | Socratic Feb 27, 2017 · $Mg(OH)_2$ Now we know that hydroxides are salts of HO^- , and some metal ion. Now if the parent metal has an electronic configuration of 2:8:2, then there are 12 electrons, and the atomic number of the metal is equal to 12. We look on the Periodic Table, and we find that $Z=12$, for "magnesium metal". As a Group 2 metal, magnesium forms a Mg^{2+} ion, and hence its hydroxide is $Mg(OH)_2$.

Question #627ef - Socratic functional groups like -OH, -CHO, -CO- are present in carbohydrates. commonly two functional are present in a carbohydrate, -OH and -CHO, or -OH and -CO-. Aldose shows different reactions and test like Tollen's silver mirror test, felling's test. different types of reactions are also shown by carbohydrates like Osazone synthesis, length increasing reaction, ...

Question #44ec4 - Socratic That's a silly way to say carbon dioxide! $CO_2(g) + Ca(OH)_2(aq) \rightarrow Ca^{2+}(aq) + H_2CO_3(aq)$ May be wrong, but it seems reasonable. The gas would interact with the hydroxide and water molecules to form carbonic acid. There are also concentrations of OH^- and H^+ leftover from this reaction and from the auto-ionization of water, but that's probably going too far into detail. **If the pH=9 for a solution of KOH(aq), what is the ...** - Socratic Aug 18, 2016 · If the pH = 9 for a solution of $KOH(aq)$, what is the concentration of H^+ ? **What are the spectator ions in the reaction between sodium** Oct 16, 2016 · The nitrate and the sodium ions. $Na_2CO_3(aq) + 2AgNO_3(aq) \rightarrow Ag_2CO_3(s) + 2NaNO_3(aq)$ The net ionic equation is: $2Ag^+(aq) + CO_3^{2-}(aq) \rightarrow Ag_2CO_3(s)$

Ag_2CO_3(s)darr ... **What makes for a good leaving group? + Example - Socratic** Jun 23, 2015 · A good leaving group has to be able to part with its electrons easily enough, so typically, it must be a strong acid or weak base relative to other substituents on the same molecule. It helps to know the pKa of what would be leaving. Let's say you had a mechanism where you are trying to do an E2 reaction to make an -OH (hydroxyl) group leave. Maybe you have this compound on hand, sec-butanol ...

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