

# John Birmingham Axis Of Time

M Carnoy

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heat for water 4 184 j  $q_{\text{oc}} = 1$  what is the specific heat of aluminum if the temperature of a 28 4 g sample of aluminum is increased by 8 1 oc when 207 j of heat is added 2

### 7 3 heats of reactions and calorimetry chemistry

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the final temperature is exactly halfway between  $t_1$  and  $t_2$  the final temperature is closer to  $t_2$  than to  $t_1$  c

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