

# Income Tax Shattering The Myth

Jianjun Gao

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 $2 \text{ CO}_2 \text{ g} + \text{H}_2\text{O g} \rightarrow \text{C}_2\text{H}_2 \text{ g} + \frac{5}{2} \text{O}_2 \text{ g} + 2 \text{H}_2 \text{ g} + \text{C}_2\text{H}_6 \text{ g} \quad \Delta h = 94.5 \text{ kJ}$   
 $\text{H}_2\text{O g} + \text{H}_2 \text{ g} + \frac{1}{2} \text{O}_2 \text{ g} \rightarrow \text{H}_2\text{O l} \quad \Delta h = 71.2 \text{ kJ}$   
 $\text{C}_2\text{H}_6 \text{ g} + 2 \text{CO}_2 \text{ g} + 3 \text{H}_2 \text{O g} \rightarrow \text{C}_2\text{H}_4 \text{ g} + \text{H}_2 \text{ g} + 2 \text{CO g} + 3 \text{H}_2 \text{O l} \quad \Delta h = 283 \text{ kJ}$

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web chemistry 120 hess s law worksheet calculate  $\Delta h$  for the reaction  
 $\text{C}_2\text{H}_4 \text{ g} + \text{H}_2 \text{ g} \rightarrow \text{C}_2\text{H}_6 \text{ g}$   
 from the following data  
 $\text{C}_2\text{H}_4 \text{ g} + 3 \text{O}_2 \text{ g} \rightarrow 2 \text{CO}_2 \text{ g} + 2 \text{H}_2\text{O l} \quad \Delta h = 1411 \text{ kJ}$   
 $\text{C}_2\text{H}_6 \text{ g} + 7 \text{O}_2 \text{ g} \rightarrow 2 \text{CO}_2 \text{ g} + 3 \text{H}_2\text{O l} \quad \Delta h = 1560 \text{ kJ}$

285 kj mole

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web thus for reaction 9 2 in which 1 mole of  $\text{H}_2\text{O}$  and 1 mole of  $\text{OH}^-$  combine to form 2 moles of  $\text{H}_2\text{O}$  h 55 8 kj as shown note that the enthalpy change  $\Delta h_{\text{products}} - \Delta h_{\text{reactants}}$  is positive if heat is absorbed that is if  $\Delta h_{\text{products}}$

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reactions in this lab three experiments were conducted in order to test hess s law the experiments required observing and collecting data on three different reactions

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web 3 12 hess s law in this experiment you will measure the amount of heat released in these three related exothermic reactions  $\text{NaOH} + \text{HCl} \rightarrow \text{NaCl} + \text{H}_2\text{O}$   $\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow \text{Na}_2\text{SO}_4 + \text{H}_2\text{O}$   $\text{NaOH} + \text{H}_3\text{PO}_4 \rightarrow \text{Na}_3\text{PO}_4 + \text{H}_2\text{O}$

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najimy wendy makkena mary wickes and harvey keitel

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