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J Elliott

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web the pcl 3 lewis structure has the typical case of
phosphorus p in the center with 3 bonds to 3 other atoms
phosphorus is from the same column as nitrogen in the
periodic table meaning that p and n generally have the same
bonding structure note the lone pair dots without bonds on
top of p just like for n in the previous example for nh 3
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have based on the number of valence electrons in each atom
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