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which is heaviest a 25 g of hg b 2 mole

of h₂o c 2 mole of co₂ d 4 g atom of o q

2 16 g of sox occupies 5 6 litre at stp

assuming ideal gas nature the value of

x is a 1 b 2 c 3 d none of these q 3 the

density of liquid mol wt 70 is 1 2 g ml 1

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02x10²³ 5 hint first find the number of

moles of oxygen used 1027 8g 1023 2g

4 6g o₂ 2 16 32 g mol n m mr n 4 6 32

n 0 144 mol y n l y 0 144 6 02 1023 y 8

669 molecules

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unbelievably small atoms and

molecules that make something up to

the measurable properties such as

mass which we may observe in a

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understanding of the mole concept is

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






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